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## Chromatofocusing of human hemoglobins

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Chromatofocusing is a relatively new technique in which a pH gradient is produced on an ion exchanger by utilizing the buffering action of the charged groups on the ion exchanger [1, 2]. This article describes separation of the commonly occurring human hemoglobins A<sub>2</sub>, E, C, S, A<sub>0</sub>, F<sub>0</sub>, and F<sub>1</sub> by this chromatographic technique.

## EXPERIMENTAL

### *Reagents*

Polybuffer 96, polybuffer exchanger PBE 94, and columns were obtained from Pharmacia Fine Chemicals (Piscataway, NJ, U.S.A.). All other chemicals were commercial analytical grade products.

### *Hemoglobin preparations*

Human red blood cells were collected from whole blood in EDTA anticoagulant by centrifugation and washed three times with 5 volumes of 0.15 M sodium chloride. Cord blood cells were the source for fetal hemoglobin. The washed red cells were lysed with 2 volumes of deionized water.

### *Hemoglobin electrophoresis*

Hemoglobin electrophoresis was performed on cellulose acetate at pH 8.4 and on citrate agar at pH 6.2 with an apparatus from Helena Laboratories (Beaumont, TX, U.S.A.) [3].

### *Chromatofocusing*

Chromatofocusing was performed with 0.9 cm diameter columns containing

21–54 cm of anion-exchanger gel PBE 94. The eluates were continuously monitored (percent transmission) at 540 nm with a Model 100-10 spectrophotometer and flow-through cuvette (Hitachi, Tokyo, Japan) that was interfaced with a strip chart recorder (Linear Instruments, Irvine, CA, U.S.A.).

The anion exchanger was equilibrated with starting buffer, 0.025 *M* imidazole · HCl, pH 7.4 or pH 8.0, by passing several column bed volumes of buffer through the column. The eluting ampholyte polybuffer, PBE 96, was prepared by diluting the stock with 10 volumes of deionized water, adjusting the pH to 6.0 with 1 *M* hydrochloric acid, and adding water to a final dilution of 1:13. All buffer solutions were degassed under vacuum (water aspirator) immediately prior to use.

A mixture of hemoglobin variants was prepared by combining erythrocyte lysates that contained hemoglobins E, C, A<sub>2</sub>, S, A<sub>0</sub>, F<sub>0</sub>, and F<sub>1</sub>. One hundred microliters of this mixture containing 4–5 g/dl hemoglobin were applied to the column. Elution with polybuffer 96 was performed by gravity flow at a rate of 14–24 ml/h. Electrophoretic analysis of the fractions was performed after concentrating them in an A25 Amicon macrosolute concentrator (Amicon, Lexington, MA, U.S.A.).

## RESULTS

With a 54 × 0.9 cm column excellent separations of hemoglobins E plus A<sub>2</sub>, S, A<sub>0</sub>, F<sub>0</sub>, F<sub>1</sub> were achieved within 18 h, but hemoglobins A<sub>2</sub> and E failed to separate (Fig. 1). Hemoglobin C was not available for this separation. The hemoglobin type in each peak was confirmed by electrophoresis, although neither chromatofocusing nor electrophoresis resolved hemoglobins E and A<sub>2</sub>. Hemoglobins F<sub>0</sub> and F<sub>1</sub> were further identified by their resistance to denaturation in alkali [5, 6].

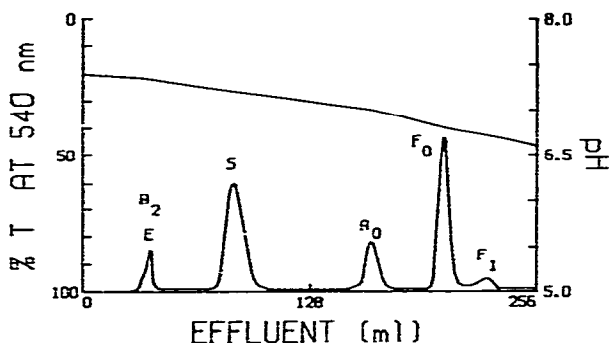


Fig. 1. Chromatofocusing of human hemoglobin variants A<sub>2</sub>, E, S, A<sub>0</sub>, F<sub>0</sub>, and F<sub>1</sub> on a long column. A mixture of oxyhemoglobins (A<sub>2</sub>, E, S, A<sub>0</sub>, F<sub>0</sub>, F<sub>1</sub>) containing a total concentration of 4.7 mg hemoglobin was chromatographed on a PBE gel column, 54 × 0.9 cm, starting at pH 7.4, 0.025 *M* imidazole · HCl and eluting by gravity with pH 6.0 polybuffer 96–HCl at a flow-rate of 14.3 ml/h.

A similar pattern (Fig. 2) with high resolution can be achieved in about 6 h using a shorter column (26 × 0.9 cm). Hemoglobins S, A<sub>0</sub>, F<sub>0</sub> and F<sub>1</sub> were clearly resolved from each other and from a heterogeneous peak that consisted

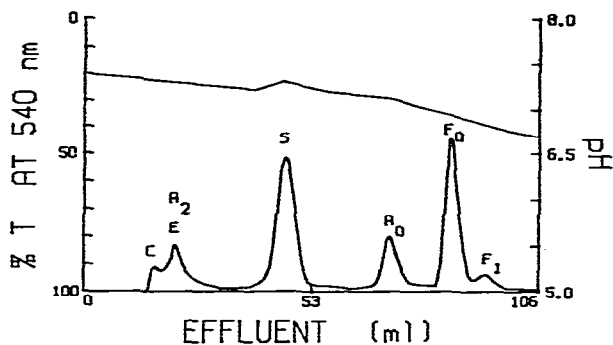


Fig. 2. Chromatofocusing of human hemoglobin variants C, A<sub>2</sub>, E, S, A<sub>0</sub>, F<sub>0</sub> and F<sub>1</sub> on a short column (pH 7.4 to 6.0). Conditions were the same as in Fig. 1 except that the column dimensions were 26 × 0.9 cm with a flow-rate of 17.5 ml/h and hemoglobin C was added to the mixture.

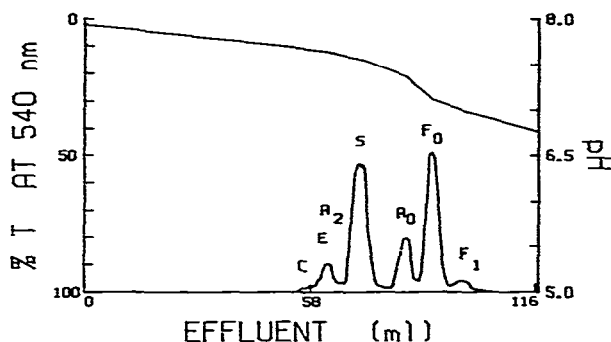


Fig. 3. Chromatofocusing of human hemoglobin variants C, A<sub>2</sub>, E, S, A<sub>0</sub>, F<sub>0</sub>, and F<sub>1</sub> on a short column (pH 8.1 to 6.0). Conditions were the same as in Fig. 1 except that the column dimensions were 21 × 0.9 cm with a flow-rate of 23.6 ml/h and the starting buffer was at pH 8.1.

of hemoglobins A<sub>2</sub>, E and C, although C migrated slightly faster than A<sub>2</sub> and E.

When the elution was performed with pH 8.1 rather than pH 7.5 polybuffer, resolution of the hemoglobin variants was less satisfactory and elution of the peaks occurred later (Fig. 3).

A comparison of the isoelectric points of the hemoglobin variants and the pH at which they chromatofocused showed that the heme proteins emerged at a pH close to their *pI*, and in the order of their isoelectric points, except that hemoglobin F<sub>0</sub> emerged after A<sub>0</sub> even though F<sub>0</sub> has a more basic *pI* (Table I). Moreover, we observed that the pH at which the hemoglobins focused varied with the pH of the starting buffer. When the starting pH was at 8.0 the more basic hemoglobins A<sub>2</sub>, C, E, S and A<sub>0</sub> eluted slightly above their *pI* values, while F<sub>0</sub> and F<sub>1</sub> eluted near their *pI* values, whereas with a starting buffer at pH 7.4, hemoglobins A<sub>2</sub>, E, C, S, F<sub>0</sub> and F<sub>1</sub> eluted near their *pI* values, while A<sub>0</sub> emerged from the column at a pH above its *pI*.

Attempts to separate hemoglobins C, A<sub>2</sub> and E by starting at pH 7.8 and eluting with either pH 7.2 or pH 6.8 failed to resolve these heme protein variants.

TABLE I

## ISOELECTRIC POINTS OF HEMOGLOBIN VARIANTS COMPARED WITH ELUTION pH IN CHROMATOFOCUSING

Hemoglobin species	pI*	Elution pH**	
		Gradient pH 7.4 to 6.0	Gradient pH 8.1 to 6.0
C	7.40	7.36	7.63
E	7.40	7.34	7.63
A <sub>2</sub>	7.40	7.34	7.63
S	7.25	7.26	7.55
A <sub>0</sub>	6.95	7.06	7.36
F <sub>0</sub>	7.15	6.86	7.11
F <sub>1</sub>	6.90	6.74	6.98

\*Obtained by isoelectric focusing in polyacrylamide gel [4] except for hemoglobin E which was estimated from our data.

\*\*From Figs. 1-3.

## DISCUSSION

Protein purification by chromatofocusing is dependent on eluting a given protein at, or near, its pI [1, 2], and thus, the choice of a narrow pH interval from 7.4 to 6.0 was sufficient to separate most of the commonly occurring human hemoglobins as sharp peaks with 0.05 pH units at half width. These hemeproteins emerged from an anion-exchange column in order of their isoelectric points, with the exception of fetal hemoglobin, which behaved as if it has a more acidic pI, a phenomenon that was also observed with pH-gradient chromatography on DEAE-Sephadex [7].

By using a pH interval of 8.1 to 6.0 for chromatofocusing, the hemoglobins emerged from the column at pH values somewhat higher than their pI (except for hemoglobin F<sub>0</sub>). This was probably due to some separation by ampholyte displacement because the hemeproteins were eluted after 4 to 8 column bed volumes of eluents had passed through the column (Fig. 3) when ambient buffer concentration increased and ampholyte displacement was more pronounced [8].

Others have utilized ampholyte displacement chromatography to separate human hemoglobin variants [9]. Mixtures of A<sub>2</sub> and A<sub>0</sub>, A<sub>2</sub>, S and A<sub>0</sub>, and C, S, and F were resolved but no experiments with a complete mixture of all the variants were reported. Although ampholyte displacement chromatography is an effective procedure for separating some hemoglobin variants, its major limitation is the high cost of the carrier amphoteric buffers which are used in many-fold greater concentrations than in chromatofocusing [8, 9].

The method described in this report is especially useful for preparative isolation of hemoglobin species. But its application as a routine analytical procedure would not appear to be as convenient as gel electrophoresis which is faster and less expensive than chromatofocusing.

## REFERENCES

- 1 L.A.AE. Sluyterman and O. Elgersma, *J. Chromatogr.*, 150 (1978) 17.
- 2 L.A.AE. Sluyterman and J. Wijdenes, *J. Chromatogr.*, 150 (1978) 31.
- 3 R.M. Schmidt and S. Holland, *Clin. Chem.*, 20 (1974) 591.
- 4 J.W. Drysdale, P. Righetti and H.F. Bunn, *Biochim. Biophys. Acta*, 229 (1971) 42.
- 5 D.P. Molden, N.M. Alexander and W.E. Neeley, *Amer. J. Clin. Path.*, 77 (1982) in press.
- 6 W.E. Neeley and N.M. Alexander, unpublished results.
- 7 T.H.J. Huisman and A.M. Dozy, *J. Chromatogr.*, 19 (1965) 160.
- 8 *Chromatofocusing Handbook*, Pharmacia Fine Chemicals, 1981.
- 9 C. Chapuis-Cellier, A. Francina and P. Arnaud, *Protides Biol. Fluids*, 27 (1979) 743.